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point is connected with the other points to form a specific body. The basis for each point should be: 1 - Symmetrical parts in terms of structure, arrangement and direction. 2.

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Chapter 1: Crystal Structure Asst. Prof. Dr. Tagreed

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The focus of this chapter is on one-dimensional periodic potentials, with traditional contributions going back to the early times of the foundation of quantum mechanics; brief digressions are given to evidence the conceptual link between this historical area of solid state and other much younger and very active areas of research such as artificial structures and superlattices, photonic crystals, atomic condensates.

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Charles Kittel New edition of the most widely-used textbook on solid state physics in the world. Describes how the excitations and imperfections of actual solids can be understood with simple models that have firmly established scope and power. The foundation of this book is based on experiment, application and theory.

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If NaCl is doped with 10 – 4 mol % of Sr Cl 2, the
concentration of cation vacancies will be ($N_A = 6.02$
 $\times 10^{23} \text{ mol}^{-1}$) [CBSE AIPMT 2007] 6.02×10^{16}
 mol^{-1}

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Metallic solid conducts electricity in solid state but

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ionic solids do so only in molten state or in solution or metals conduct electricity through electrons and ionic substances through ions. Metallic solids are malleable and ductile while ionic solids are hard and brittle.

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Chemistry- Solid state class 12 full chapter (Hindi ...

Ans: It is a covalent or network solid. 1.8. Why are ionic solids conducting in the molten state and not in the solid state ? Ans: In the ionic solids, the electrical conductivity is due to the movement of the ions. Since the ionic mobility is negligible in the solid state, these are non-conducting in this state.

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