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E1-26 The room volume is $(21 \times 13 \times 12) \text{ft}^3 (0.3048 \text{ m/ft})^3 = 92.8 \text{m}^3$.

The mass contained in the room is $(92.8 \text{m}^3)(1.21 \text{kg/m}^3) = 112 \text{kg}$:

E1-27 One mole of sugar cubes would have a volume of $N_A \times 1.0$

cm^3 , where N_A is the Avogadro constant. Since the volume of a

cube is equal to the length cubed, $V = l^3$, then $l = \sqrt[3]{N_A \text{ cm}^3} = 8.4$

10^7 cm . E1-28 The number of seconds in a week is

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