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WORKSHEET 16 Section 11.2 Determine the mole ratios for each of the following balanced chemical equations. $1.2C(s) + O_2(g) \rightarrow 2CO(g)$

$1 \text{ mol } O_2 : 2 \text{ mol } C : 2 \text{ mol } CO$
 $2 \text{ mol } C : 1 \text{ mol } O_2 : 2 \text{ mol } CO$
 $2 \text{ mol } CO : 1 \text{ mol } O_2 : 2 \text{ mol } C$
 $2 \text{ mol } CO : 1 \text{ mol } O_2 : 2 \text{ mol } C$

$WO_3(s) + 3H_2(g) \rightarrow W(s) + 3H_2O(l)$

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