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7. The chemical
formula for an ionic
compound
represents one
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Simplest ratio of the compound's positive ions (cations) and its negative ions (anions).

Monatomic cations are identified simply by the element's name.

Monatomic anions more complicated.

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Chemistry reveals

number of atoms

contained in a

single molecule of a

certain compound.

monoatomic ions.

ions formed from a

single atom;

remember to

change nonmetals

to -ide ending.

oxyanions.

polyatomic ions that

contain oxygen .

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binary ionic

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compounds of 2
elements between
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binary compound.

nomenclature.

oxyanion. an ion

formed from a

single atom. a

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composed of two

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system of naming
used in chemistry. a
polyatomic ion
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oxyanions. ions

formed from a

single atom.

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elements. naming

system;... name of

the cation is given

first, followed by...

polyatomic ions that

contain oxygen;...

an anion that has

one few...

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ions formed from a
single atom. binary
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Describe the contributions made by the following scientists to the development of the periodic table: a.

Stanislao

Cannizzaro b. Dmitri

Mendeleev c. Henry

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00:52. Problem 2

State the periodic

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ANSWERS Answer

the following

questions in the

space provided. 1.

Label each of the

following

statements as True

or False: True a. If

the formula mass of

one molecule is x

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amu, the molar

mass is x g/mol.
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False b.

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