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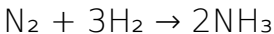
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10 from 4-5 PM PT.

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STUDY. PLAY. What
is 1 mole equal to?
 6.022×10^{23}

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representative
particles, the
amount of grams of
a substance in 1
mole. molar mass.



What is the mole
ratio between N_2
and 2NH_3 ? 1:2.

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REVIEW SHEET Ms.

Kim; Spring

Semester 2017

Period 3&5 Final

Exam: Thursday,

June 15 Period 2&4

Final Exam: Friday,

June 16 The exam

covers Chapters 8

through 14. Bring a

pencil/eraser and

calculator to the

final. You will be

provided with a

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periodic table and
other reference
needed for the
exam.

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SHEET NOTE:
Below are concepts
that we have

covered in class
throughout the
second semester.

Questions are
organized by
chapter/concept to
help you study. You
need to be able to
understand the
concepts listed

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CHEMISTRY CP

Name: Period:

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STRATEGY : Start
by reading through
your notes to
refresh your
memory on these
topics. Then, use
this review sheet

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as a starting point
to identify the
areas on which you
need to spend
more study time.

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Name _____ 37.

300.0 mL of gas has a pressure of 75.0 kPa. When the volume is

decreased to 125.0 mL, what is the pressure? 38. 50.0

L of a gases temperature is 75 °C. What is the temperature when the volume changes to 115.0

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If a chemist
calculates the
maximum amount
of product that
could be obtained
in a chemical
reaction, he or she
is calculating the ...

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CHEMISTRY I
HONORS -FINAL

EXAM REVIEW

STRATEGY: Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the

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areas on which you need to spend more study time.

For those areas, go back to homework assignments, quizzes, and reviews to practice more problems.

CHEMISTRY I
HONORS FINAL
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Name: Sonia Sung

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Date: _____

Class/Period: _____

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Sheet Directions:

Read each of the
following

questions. Answer
in the space

provided. If a
calculation is

required, SHOW

ALL WORK TO

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RECEIVE CREDIT
FOR THE ANSWER.

On the day of the
final exam, bring a
#2 pencil,
calculator, and
paper.

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have selected your
answer to each
question, blacken
the corresponding
space on the
answer sheet using
a soft, #2 pencil.
Make a heavy, full
mark, but no stray

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marks. If you decide to change an answer, erase the unwanted mark completely.

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