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____T____ 2. A covalent bond is formed by a shared pair of ...

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7-2 Review and Reinforcement (continued) Write Lewis structures for each of the following molecules. Indicated the bonds with either dots or dashes. 15. NH_3 18. PCl_3 . 16. H_2 19. CCl_4 . 17. C_2H_4 20. C_2H_6 . Answer the following questions as directed. 21. Explain why the molecule SF_4 is an exception to the octet rule.\ 7-2 Review and Reinforcement - mvhs-fuhsd.org

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Sunlevel 5-2 Review and Reinforcement (continued) Identify each of the following elements as a metal (M), nonmetal (NM), or semimetal (SM). M 19. sodium SM 20. silicon NM 21. neon M 22. calcium NM 23. nitrogen Write the family names that have been given to each of the following groups.

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Wave Properties Review from Glencoe Reinforcement S8P4f. Diagram the parts of the wave and explain how the parts are affected by changes in amplitude and pitch 2. How do changes in one part of a wave affect other parts of a wave? TEXTBOOK PAGES 466-470
*Essential Vocabulary listed in the Standards **Supplemental Vocabulary listed in the

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